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A LUPEOL DERIVATIVE FROM SALVIA PRATENSIS

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Key Word Index—Salvia pratensis; Labiatae; triterpenoids; lupeol derivative; 7β -hydroxylup-20(29)-en-3-one.

Abstract—The aerial parts of *Salvia pratensis* afforded, in addition to the known triterpenoids β -amyrin, germanicol, lupeol and loranthol, a new related lupenol, 7β -hydroxylup-20(29)-en-3-one, whose structure was elucidated by spectroscopic methods and chemical transformations.

INTRODUCTION

A reinvestigation of the chemical constituents of *Salvia* pratensis L. [1, 2] has led to the isolation of one new derivative of lupeol, 7β -hydroxylup-20(29)-en-3-one (1). Furthermore the known compounds β -amyrin, germanicol, lupeol [3] and loranthol [4] were isolated.

RESULTS AND DISCUSSION

Chromatography of the neutral fraction from an extract of Salvia pratensis afforded a crystalline dextrorotatory material, which was identified as a mixture of β -amyrin, germanicol and lupeol by $^{13}\text{C NMR}$ spectral analysis which gave data coincident with those described by Bhattacharyya et al. [3].

Mass spectroscopy (M $^+$ at m/z 440) established the molecular formula of compound 1 as $C_{30}H_{48}O_2$. Its IR spectrum indicated a hydroxyl group (3.450 cm $^{-1}$), a carbonyl group (1.705 cm $^{-1}$) and an exocyclic methylene (1.645 and 885 cm $^{-1}$).

The ¹H NMR spectrum (Table 1) of 1 showed signals for six tertiary methyl groups, a vinylic methyl (δ 1.64) which was shown to be coupled to two vinylic protons (δ 4.65 and 4.53), thus indicating the presence of an isopropenyl group and also signals for a hydroxymethine group (δ 3.83, 1H, dd, J_1 = 6.8 Hz, J_2 = 8.8 Hz) which must be axial and placed between a tetrasubtituted sp³ carbon atom and a methylene grouping.

These data suggest a 20(29)-lupene structure with one β -hydroxy group and one keto group for the triterpenoid 1. The functional groups were confirmed by the following reactions of this compound: the acetylation afforded a monoacetyl derivative 1a and the Sarett oxidation gave the diketone 3.

The carbonyl group of the compound 1 may be in any position with the exclusion of ring E (owing to its IR absorption), but it was located in position C-3 (the most likely on biosynthetic grounds) by comparison of the 1 H NMR data for H-3 in lupeol and acetoxylupeol at δ 3.20 and 4.60, respectively [5, 6], with the 1 H NMR

	\mathbb{R}^1	R ²
1	O	α- Η ; β-ΟΗ
1a	O	$\alpha \cdot H; \beta \cdot OAc$
2	α-H; β- OH	α·Η; β·ΟΗ
2a	α-Η; β -ΟΑc	α·Η; β· OAc
3	O	O

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signals corresponding to the hydroxymethine group at δ 3.83 for 1 and to the acetoxymethine group at δ 5.20 in the compound 1a.

The 13 C NMR spectrum (Table 2) of 1 was in complete agreement with the existence of a C-3 ketone [7, 8] instead of the C-3 β hydroxyl group; in particular the C-1, C-2, C-4, C-23 and C-24 carbon atom resonances (δ 39.5, 34.1, 46.8, 26.6 and 21.0, respectively) were in agreement with a 3-keto partial structure and not with those of a 3 β -hydroxy moiety [7].

The location of the hydroxyl group was limited to position C-7, 15, or 16 by the following considerations. Absorption maxima in the IR and UV spectra of the diketone 3 (v_{max} 1.705 cm⁻¹; $\lambda_{max} = 290$ nm) excluded the hydroxyl group from the E-ring or a 1,3-diketone. Position C-16 can similarly be excluded by comparison of the ¹³C NMR data of the known lup-20(29)-ene-3 β ,16 β -diol [7] with loranthol (Table 2). In the Sarett oxidation of loranthol and oxidation of 1, the same crystalline product was obtained, mp and MS (Experimental) and ¹H NMR (Table 1) were identical with the known 3,7-lupendione (3), the ¹³C NMR data are shown in Table 2. All these data give evidence that the structure of 1 is 7β -hydroxylup-20(29)-en-3-one.

EXPERIMENTAL

Mps: uncorr. Optical rotations: MeOH. UV: EtOH. IR: KBr. ¹H NMR: 200 MHz, CDCl₃, TMS as int. standard. ¹³C NMR: 50.3 MHz. MS: 70 eV. Analytical TLC: Silica gel G. Prep. TLC: Silica gel PF²³⁴⁻²³⁶. CC: Silica gel 60 (0.063=0.2 mm).

Extraction and separation of compounds. Salvia pratensis L. collected in Valgañon (La Rioja, Spain) in July 1987, voucher specimens are deposited in the Botany Department (register number SALA No. 7320). Air-dried plant material (1.4 kg), was extracted with hexane in a Soxhlet for 15 hr. and yielded 82 g of extract (5.85%). The neutral fraction (5.1 g) of the MeOH-urea (1:4) soluble extract was chromatographed over silica gel developed with hexane and hexane-Et₂O mixtures of increasing

Table 2. 13C NMR spectral data of compound 1, 1a, 2, 2a and 3

C	1	la	2	2a	3
1	39.5	39.3	38.7	38.3	39.2
2	34.1	34.0	27.5	23.7	34.0
3	217.0	216.4	78.9	80.6	215.8
4	46.8	47.0	37.3	38.1	47.4
5	52.4	51.8	52.5	52.2	53.8
6	30.0	26.6	27.5	23.7	27.8
7	74.1	76.5	27.3 74.7	77.1	213.7
8	74.1 44.5	45.6	74.7 46.9	45.6	43.7
9					
-	50.0	50.0	50.5	49.8	52.0
10	37.0	36.9	37.3	36.6	37.3
11	21.6	21.5	20.9	20.5	21.8
12	25.4	25.3	25.3	25.1	25.8
13	38.8	38.8	38.7	37.2	39.9
14	42.8	44.4	42.8	44.2	43.3
15	30.5	29.8	29.4	29.9	27.7
16	36.1	36.2	36.1	37.1	35.8
17	42.8	42.8	42.8	42.7	42.7
18	48.2	48.3	48.3	50.3	48.0
19	46.8	48.3	48.2	48.3	47.8
20	150.9	150.9	151.0	150.8	150.9
21	30.0	30.1	30.0	30.0	30.0
22	40.2	40.2	40.2	40.1	40.3
23	26.6	26.7	28.0	25.2	26.0
24	21.0	22.0	15.4	15.8	20.9
25	15.6	15.8	15.1	15.6	15.6
26	10.0	11.3	10.2	11.5	15.6
27	14.9	15.0	15.8	14.9	15.6
28	18.0	18.1	17.9	18.0	18.3
29	109.4	109.6	109.4	109.4	109.6
30	19.5	19.5	19.4	19.4	19.6
OAc		170.5		170.7	
OAc		_		170.4	
OAc		21.0		21.2	
OAc		_		20.7	

Table 1. 1H NMR data of compounds 1, 2a and 3

Н	1	1a	2a	3
H-3			4.45 dd	
			(4, 11.5)	
H-7	3.83 dd	5.20 dd	5.00 dd	
	(6.8, 8.8)	(7.0, 8.9)	(6.0, 12.0)	
H-29	4.65 dd	4.68 dd	4.70 dd	4.68 br s
	(2.2, 0.7)	(2.1, 0.7)	(2.0, 0.7)	$(W_{1/2} 4.6)$
H-29	4.53 br s	4.58 br s	4.60 dd	4.58 br s
	$(W_{1/2} \ 3.7)$	$(W_{1/2} 3.7)$	$(W_{1/2} 3.7)$	$(W_{1/2} \ 3.6)$
C(Me 30)	1.64 d	1.67 d	1.64 d	1.68 d
	(0.7)	(0.7)	(0.7)	(0.7)
	1.06 s	$1.28 \ s$	1.13 s	1.33 s
	1.04 s	1.20 s	0.98 s	1.15 s
	1.01 s	1.07 s	$0.83 \ s$	1.06 s
	0.99 s	1.02 s	$0.83 \ s$	1.03 s
	0.89 s	0.92 s	$0.79 \ s$	1.03 s
	0.77 s	$0.80 \ s$	0.76 s	$0.80 \ s$
OAc		2.00 s	2.03 s	
			1.97 s	

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polarity to give a crystalline mixture of β -amyrin, germanicol and lupeol (710 mg, hexane-Et₂O, 9:1); 1 (1.05 g, hexane-Et₂O, 4:1) and loranthol (2) (431 mg, hexane-Et₂O, 7:3). The different components were purified by crystallization.

 7β -Hydroxylup-20(29)-en-3-one (1). Colourless crystals, mp 217 219° (MeOH). IR $v_{\rm max}$ cm⁻¹: 3450, 3060, 1705, 1645 and 885. ¹H and ¹³C NMR in Tables 1 and 2. MS m/z (rel. int.): 440 (21), 425 (14), 399 (8), 384 (12), 371 (12), 303 (10), 387 (9), 257 (8), 235 (23), 205 (11), 183 (25), 155 (22), 137 (10), 69 (100), 41 (23).

$$\frac{\lambda}{\alpha} = \frac{589}{+10.9} = \frac{578}{+13.6} = \frac{546}{+26.5} = \frac{436}{+26.5} = \frac{$$

Acetylation of compound 1. Treatment of a crystalline sample of compound 1 with Ac_2O pyridine in the usual way afforded the monoacetate 1a. IR $\nu_{\rm max}$ cm⁻¹: 3060, 1730, 1705, 1645, 1220 and 885. ¹H and ¹³C NMR in Tables 1 and 2.

Oxidation of compound 1. A mixture 40 mg or 1 was dissolved in pyridine (0.2 ml) and added to a soln of CrO_3 (32 mg) in pyridine (0.2 ml) and CH_2Cl_2 (1 ml). The mixture was stirred for 4 hr in an ice water bath under N_2 . The oxidation product was chromatographed over silica gel. Elution with hexane– Et_2O (19:1) afforded 32 mg of lup-20(29)-en-3,7-dione (3). Colourless crystals, mp 203–205° (MeOH). MS m/z (rel. int.): 438 (78), 423 (14), 370 (12), 327 (59), 247 (56), 234 (100), 205 (62), 130 (36), 109 (54), 95 (49), 41 (26).

Loranthol (2). Colourless crystals, mp 223–225° (MeOH). MS *m/z* (rel. int.): 442 (100), 427 (30), 409 (10), 391 (8), 386 (12), 332 (51), 331 (62), 249 (33), 236 (63), 223 (22), 217 (56), 203 (4), 195 (13), 189 (8), 175 (9), 139 (11).

$$\frac{\lambda}{[\alpha]} = \frac{589}{+12} = \frac{578}{+14} = \frac{546}{+16.5} = \frac{436}{+32.5}$$
 (c 0.4).

Acetylation of Loranthol (2) afforded the diacetate 2a. Colourless crystals mp 216° (CH₂Cl₂-hexane). IR v_{max} cm⁻¹: 3065, 1740, 1730, 1650, 1225, 890. ¹H and ¹³C NMR in Tables 1 and 2.

$$\frac{\lambda}{[\alpha]} \frac{589}{+21} \frac{578}{+22.3} \frac{546}{+25.7} \frac{436}{+44.8} (c \ 0.4)$$

Oxidation of loranthol (2). Sarett oxidation of 2 afforded lup-20(29)-en-3,7-dione (3).

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